## collisions Higher education

# CONTENT AREA OVERVIEW Lewis Structures



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#### Challenges

#### Sandbox

- The Challenge Levels increase in rigor and complexity.
- The first 7 levels are tutorial levels.
  - 17 core levels
  - 4 Connected Levels to Radii Trends
- The Sandbox is an exploratory learning space for extended practice and review of Lewis Structures.
- 14 Achievements

#### Integrated Chemistry Concepts

- Single Bonds, Double Bonds, and Triple Bonds
- Octet and Duet Rules
- Electron Domains
- Electronegativity
- Bond Polarity
- VSEPR Theory and Molecular Geometry
- Coordinate Covalent Bonds

GAMEPLAY BASICS





#### Achievements



#### **Selected Bank of Atoms**

The bank includes the following atoms:

HydrogenPhosphorusCarbonSulfurNitrogenChlorineOxygenSeleniumFluorineBromineSiliconSilicon





LEVELS 1-17 GOAL: Bond the atoms to complete the shapes and hit the bond polarity targets.



Lewis Structures Challenges LEVEL TARGETS

#### RADII TRENDS TO LEWIS STRUCTURES CONNECTED LEVELS GOAL:

Some atoms are missing from the bank. Use the button on the left to go to Radii Trends. Solve the challenge and bring back the missing atoms!





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## **CHEMISTRY CONNECTIONS**





VALENCE ELECTRONS REPRESENTED

OCTET SATISFIED







SINGLE BOND 2 SHARED ELECTRONS



DOUBLE BOND 4 SHARED ELECTRONS



TRIPLE BOND 6 SHARED ELECTRONS



## **CHEMISTRY CONNECTIONS**



#### CHEMISTRY CONCEPT: Bond polarity

Players can create non-polar, semi-polar, and very polar bonds by combining atoms of various electronegativities.



PLAYER CAN OBSERVE SHARED ELECTRONS MOVING CLOSER TO THE ATOM WITH THE HIGHER ELECTRONEGATIVITY.





## **CHEMISTRY CONNECTIONS**



BENT 120°



the basic molecular shapes with 4 or fewer electron domains.







HOINSOH H

0



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INCORRECT

CORRECT